

A NOVEL AND SIMPLE METHOD FOR THE PREPARATION OF ADENINE ARABINOSIDE BY BACTERIAL TRANSGLYCOSYLATION REACTION

T. UTAGAWA, H. MORISAWA, T. MIYOSHI, F. YOSHINAGA, A. YAMAZAKI and K. MITSUGI

Central Research Laboratories, Ajinomoto Co., 1-1 Suzuki-cho, Kawasaki-ku, Kawasaki, 210, Japan

Received 23 October 1979

Revised version received 16 November 1979

1. Introduction

The antibiotic, ara-A (adenine arabinoside, 9- β -D-arabinofuranosyladenine), was originally synthesized as a potential anticancer agent [1] and also found to possess significant activities against DNA viruses including cytomegalovirus in vitro, vaccinia virus in vivo and Epstein-Barr virus in vitro [2-5]. Ara-A is clinically effective against various manifestations of herpes simplex and herpes zoster virus infection in humans as shown [6,7]. Thus, the development of a new method for the preparation of ara-A is of considerable interest. Fermentation [8] and chemical synthesis [9-11] methods have been used for the preparation of ara-A. These methods were low yielding, laborious and time-consuming. Here, we describe a simple method for the production of ara-A by a bacterial or enzymatic arabinose-transfer reaction (transarabinosylation, fig.1).

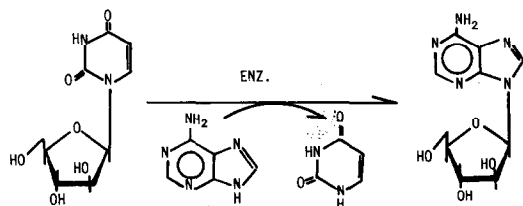


Fig.1. Transarabinosylation.

2. Materials and methods

Ara-A forming microorganisms were selected from stock cultures of our laboratory. The cells of *Enterobacter aerogenes* AJ 11125 were used as a

likely source of enzyme for transarabinosylation; they were grown in a medium containing 1% pepton, 1% meat extract, 0.5% yeast extract and 0.5% NaCl in tap water, and the medium adjusted to pH 7.0 by adding 1 N KOH. The microbial strain was cultured aerobically at 30°C for 36 h with shaking in a 500 ml Erlenmeyer flask containing 100 ml medium. Ara-U was easily prepared by reaction of uridine with ethylene carbonate followed by acidic treatment as in [12,13]. All other chemicals were commercial products. Ara-A was determined using high pressure liquid chromatography: resin, Zipax SCX; column size, 2.1 \times 500 mm; solvent, 0.05 N HNO₃; pressure, 50 kg/cm²; flowrate, 1.5 ml/min; detector, ultra-violet monitor (260 nm).

3. Results

3.1. Distribution of ara-A producer in microorganisms

Ara-A-producing activity was widely distributed in various strains of bacteria, especially in the genera *Enterobacter*, *Erwinia* and *Escherichia* (table 1) where *Enterobacter aerogenes* AJ 11125 was selected as the best producer which gave the highest activity in its transarabinosylation reaction. Intact cells (50 mg as wet paste) of this organism produced 2 mg/ml of ara-A in the standard incubation mixture.

3.2. Optimum pH

The relative activity for ara-A production by *Enterobacter aerogenes* AJ 11125 at various pH values was determined by adjusting the incubation mixture pH with 1 N KOH or HCl. The optimum pH was \sim 7.0 at 60°C as shown in fig.2.

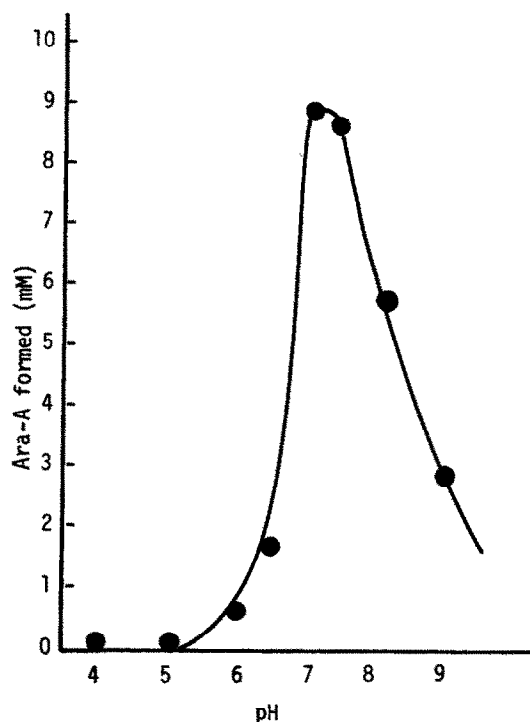


Fig. 2. Effect of pH on ara-A formation. *Enterobacter aerogenes* AJ 11125 was cultured as in the text. Each incubation mixture containing 5% intact cells as wet paste, 30 mM ara-U, 10 mM adenine and 25 mM potassium phosphate and adjusted to various pH values with 1 N KOH or HCl, was incubated for 5 h at 60°C.

3.3. Optimum temperature.

As shown in fig. 3, the optimum temperature for ara-A formation by *Enterobacter aerogenes* AJ 11125 was 60–65°C, at pH 7.0. If the reaction was at <50°C, adenine was deaminated to hypoxanthine and no ara-A was formed at ~37°C.

3.4. Isolation and identification of the product.

Intact cells (30 g as wet paste) of *Enterobacter aerogenes* AJ 11125 were suspended in the incubation mixture containing ara-U (7.3 g), adenine (1.3 g) and KH_2PO_4 (3.4 g) in 11 total vol. (pH 7.0) and incubated at 60°C for 15 h with gentle agitation. Under these conditions, 92% of the adenine on a molar basis was transformed into ara-A (2.4 g). The incubation mixture was boiled for 5 min to terminate

Table 1
Distribution of ara-A producer in microorganisms

Genera	No. strain tested	No. strain forming ara-A
<i>Pseudomonas</i>	40	1
<i>Xanthomonas</i>	6	0
<i>Agrobacterium</i>	6	0
<i>Alcaligenes</i>	6	0
<i>Escherichia</i>	33	31
<i>Citrobacter</i>	2	2
<i>Salmonella</i>	3	3
<i>Kluyvera</i>	5	5
<i>Enterobacter</i>	13	10
<i>Serratia</i>	4	3
<i>Proteus</i>	11	10
<i>Erwinia</i>	7	6
<i>Aeromonas</i>	4	3
<i>Flavobacterium</i>	31	4
<i>Achromobacter</i>	4	1
<i>Bacterium</i>	2	2
<i>Micrococcus</i>	8	0
<i>Staphylococcus</i>	2	0
<i>Microbacterium</i>	3	0
<i>Planococcus</i>	6	0
<i>Sarcina</i>	6	0
<i>Bacillus</i>	13	0
<i>Sporosarcina</i>	1	0
<i>Corynebacterium</i>	7	0
<i>Arthrobacter</i>	3	0
<i>Brevibacterium</i>	17	0

The incubation mixture containing 5% wet cells, 0.5% ara-U, 0.2% adenine and 0.5% KH_2PO_4 was adjusted to pH 7 with 1 N KOH and was incubated at 60°C for 15 h with gentle agitation. The reaction was terminated by heating at 100°C for 5 min. Bacteria which formed >10 mg/l of ara-A were judged as positive strain

the reaction, then centrifuged to remove the bacterial cells. The supernatant was kept for several hours at 4°C, and the precipitant was collected by filtration and recrystallized from water to give white needles (2.0 g).

This compound was proved to be identical with an authentic ara-A sample in all physical properties: melting point 256°C (dec.); $[\alpha]_D^{40} -5.2^\circ$ (0.25% in water); $\text{UV}^{pH 7}_{\text{max}} 259 \text{ nm}$ (ϵ 14 500), $\text{pH } 11_{\text{max}} 259 \text{ nm}$ (ϵ 14 300); NMR data (DMSO-d_6), 8.19 ppm (1 H, s, H-8), 8.25 ppm (1 H, s, H-2), 7.35 ppm (2 H, s, NH_2), 6.33 ppm (1 H, d, H-1', $J_{1'2'} 4.5 \text{ Hz}$): Elemental analysis for $\text{C}_{10}\text{H}_{13}\text{N}_5\text{O}_4$ calc. C, 44.94; H, 4.90; N, 26.21; found. C, 44.60; H, 5.01; N, 26.15.

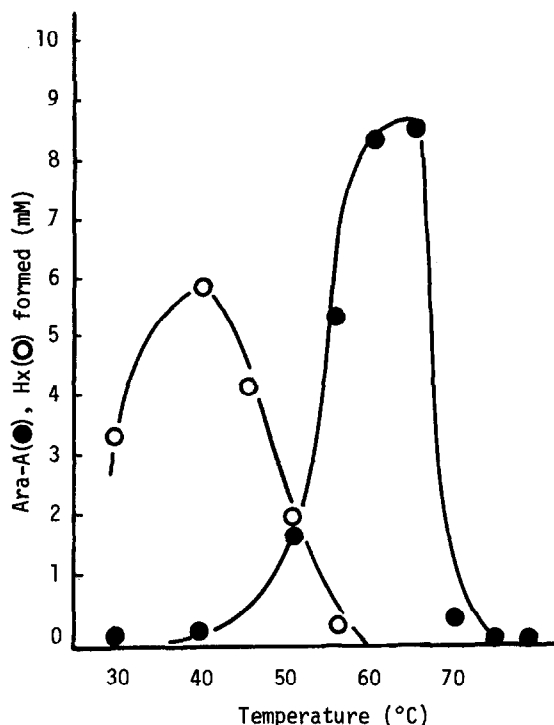


Fig.3. Effect of temperature on ara-A formation. *Enterobacter aerogenes* AJ 11125 was cultured as in the text. Each incubation mixture containing 5% intact cells as wet paste, 30 mM ara-U, 10 mM adenine and 25 mM potassium phosphate buffer (pH 7.0) was incubated at various temperatures for 5 h. Hx, hypoxanthine.

4. Discussion

Arabinose-transfer activity from ara-U to adenine is distributed in many microorganisms. In particular, *Enterobacter aerogenes* AJ 11125 was found to have the highest activity.

A chemical transribosylation reaction from pyrimidine to purine has been reported by some investigators [14,15]. The synthetic method for purine nucleosides by chemical transglycosylation, however, has inevitable defects such as the formation of α or β anomers or 7 and 9 isomers in some cases. On the contrary, the microbial or enzymatic transarabino-sylation gives stereospecifically the 9- β -D-arabino-furanosyl derivative alone.

Of special interest is the optimal temperature of the catalytic reaction. At $\sim 37^\circ\text{C}$, hypoxanthine was formed without any ara-A formation. This suggested that the adenine-deaminase activity in bacterial cells was high at 37°C . On the other hand, the deaminase

activity was completely depressed at $\sim 60\text{--}65^\circ\text{C}$, and ara-A production took place favourably.

In addition, it must be pointed out that arabinofuranose-1-phosphate isolated from the incubation mixture may be an intermediate in the formation of ara-A because no ara-A was produced in the absence of P_i (unpublished). From these results, it seems that the transarabinosyl reaction may be catalyzed by a nucleoside phosphorylase.

The present method, using a combination of chemical and enzymatic reactions, offers the most preferable route for the preparation of ara-A because of its stereospecificity and simplicity.

Acknowledgements

The authors express their deep gratitude to Dr Takekazu Akashi, Director of the Central Res. Labs., for his helpful advice and encouragement. They are also grateful to Drs T. Shiro, Y. Komachiya, T. Miyamae and Y. Hirose of Ajinomoto Co., for their support and valuable advice during this work.

References

- [1] Lee, W. W., Benitez, A., Goodman, L. and Baker, B. R. (1960) *J. Am. Chem. Soc.* 82, 2648–2649.
- [2] Sidwell, R. W., Arnett, G. and Dixon, G. J. (1967) *Abstr. Intersci. Cont. Antimicrobial Agents Chemotherapy*, 7th. Chicago, p. 28.
- [3] Sidwell, R. W., Dixon, G. J., Sellers, S. M. and Schabel, F. M. (1968) *Appl. Microbiol.* 16, 370–382.
- [4] Schabel, F. M. (1968) *Chemotherapy* 13, 321–338.
- [5] Benz, W. C., Siegel, P. J. and Baer, J. (1978) *J. Virol.* 27, 475–482.
- [6] Pavan-Langston, D., Buchanan, R. A. and Alford, C. A., jr (1975) *Adenine arabinoside; an antiviral agent*, Raven Press, New York.
- [7] Whitley, R. J., Sen-Jaw Soong, Dolin, R., Galasso, G. J., Chilen, L. T. and Alford, C. A. (1977) *New Engl. J. Med.*, 297, 289–294.
- [8] Parke, Davis and Co. (1967) *Belgian Pat.* 671, 557.
- [9] Galudemans, C. P. J. and Fletcher, H. G. (1963) *J. Org. Chem.*, 28, 3004–3006.
- [10] Ikehara, M. and Ogiso, Y. (1972) *Tetrahedron* 28, 3695–3704.
- [11] Ranganathan, R. (1975) *Tetrahedron Lett.* 1185–1188.
- [12] Komura, H., Yoshino, T. and Ishido, Y. (1973) *Bull. Chem. Soc. (Japan)* 46, 550–553.
- [13] Brown, D. M., Sir Todd, A. and Varadarajan, S. (1965) *J. Chem. Soc.* 2388–2393.
- [14] Miyaki, M., Saito, A. and Shimizu, B. (1970) *Chem. Pharm. Bull.* 18, 2468–2469.
- [15] Azuma, T. and Isono, K. (1977) *Chem. Pharm. Bull.* 25, 3347–3353.